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MOF	organic trameworks (J. Am. Chem. Soc., DOI: 10.1021/ja307953m). The two-step technique allows researchers to control the films' thickness and could							
work with many types of the framework materials, known as MOFs. The advance could make it easy for scientists to use MOFs in hydrogen storage and gas sensors, the researchers say.								
Water Thin film on water Stamping MOFs are crystalline materials that contain metal clusters connected by organic molecules. Their structure makes the materials very porous and gives them a								
FILMS MADE EASY large surface area. Because of these properties, researchers have studied   To make thin films of metal organic frameworks, them for storing hydrogen gas to fuel vehicles. Scientists are also interested in								
researchers drip a suspension of MOF flakes in an using the materials as catalysts and sensors, both of which require large								
surface of water in a beaker (left). The flakes spread out on the water's surface to produce a thin film (second from left). The researchers then use a rubber stamp to transfer the film from the beaker (second from								
right) to a qua	right) to a quartz substrate. Credit: J. Am. Chem. Soc. The best-known way to make such thin films, and to control the films' thickness, is to grow crystalline MOF layers one by one on a substrate, says Hiroshi							
time this mo	thad isn't compatible w	Kitag	gawa, a chemist at <b>Kyoto L</b>	Jniversity, in Japan. In ac	ldition to consuming			
temperatures	s and corrosive solven	ts. These conditions	would destroy the substrate	e or the growing film itself.	i as iliyii			
Kitagawa an	d his colleagues avoid	this problem by prod	ducing the MOF separately	from growing the film. First	, they prepare their			

Kitagawa and his colleagues avoid this problem by producing the MOF separately from growing the film. First, they prepare their MOF particles of choice in solution using the material's synthetic conditions. Then they disperse the MOFs on the surface of water to produce a thin film. The team uses a rubber stamp to transfer the materials to a solid substrate. Because they place the films on the substrate after they've synthesized the MOFs, they don't have to worry about the MOF production conditions destroying their growing film or substrate.

The researchers used this assembly method to make thin films of MOFs containing copper and a type of porphyrin. They first combined copper and the porphyrin compound in *N*,*N*-diethylformamide and ethanol to create MOF flakes. Using atomic force microscopy, the researchers found that the flakes were 300 to 500 nm in diameter and about 15 nm thick. They then dispersed the

flakes in ethanol or acetone using ultrasonication. Drop by drop, the team added this suspension to the surface of water in a beaker. The flakes spread out to form a flat thin sheet, and the researchers used a stamp to lift the sheet and transfer it to a solid surface. They repeated the process to stack up layers of the sheets and make a film of a desired thickness.

The method is fast, Kitagawa says: His team could stack 100 layers of MOF sheets in 10 minutes. By contrast, the traditional method took them 10 minutes to produce a single layer.

**Osama Shekhah** at the **King Abdullah University of Science and Technology**, in Saudi Arabia, says the new method is convenient, which is important to researchers working on new applications of MOFs. He thinks the technique's only drawback is that it applies only to two-dimensional, flake-shaped MOFs that can spread out into sheets on water. Kitagawa is working to apply the technique to MOFs with three-dimensional crystal structures.

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## Comments

Bharat Jogi (June 16, 2015 4:14 AM) Thanks for the useful information. Can you please tell me whether this material/chemical can be use on PET or PBT sheet, which is used for making rubber stamp. » Reply

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